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Formulas, Equations and Moles Chapter 3 Chapter 3 2 Interpreting Chemical Equations You can interpret a balanced chemical equation in many ways.

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(Modified from Exercise 14) - Gram-mole Practice Problems. 1. ... - Practice Calculating Composition From Formula. 1. SO 2. Na 2 CO 3. FeCO 3. 4. CH ...

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Stoichiometry: Chemical Arithmetic For each equation and starting amount ...

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Calculations and Chemical Equations ... $98.6\text{g} \times 1 \text{ mole}/32\text{g} = \text{moles sulfur} = 3.07 \text{ moles}$ sulfur Practice: ... Chemical Equations and the Mole

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Chemistry 110 Practice Problems: gram/mole conversions Avogadro's Number: $N_A = 6.02 \times 10^{23} \text{ =mole}^{-1}$. In 3 moles of O_2 : (a) How many O_2 molecules are there?

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Stoichiometric Calculations: Mole to Mole Calculations ... Solution: From the balanced equation you can see that 2 S react with 3 O_2